

47. The heating coil of a hot water heater has a resistance of 20Ω and operates at 210 V. If electrical energy costs $\$0.080/\text{kWh}$, what does it cost to raise the 200 kg of water in the tank from 15°C to 80°C ? (See Chapter 11.)

Solution

The kilowatt-hour is a measure of energy equal to

$$1 \text{ kWh} = (1.00 \text{ kW})(1.00 \text{ h}) = (1000 \text{ J/s})(3600 \text{ s}) = 3.60 \times 10^6 \text{ J}$$

The energy needed to raise the temperature of 200 kg of water from 15°C to 80°C is

$$Q = mc_{\text{water}}(\Delta T) = (200 \text{ kg})(4186 \text{ J/kg}\cdot^\circ\text{C})(65^\circ\text{C}) = 5.4 \times 10^7 \text{ J}$$

In units of kWh,
$$Q = (5.4 \times 10^7 \text{ J}) \left(\frac{1.00 \text{ kWh}}{3.60 \times 10^6 \text{ J}} \right) = 15 \text{ kWh}$$

and the cost of operating the heater to produce this quantity of thermal energy is

$$\text{cost} = (\text{energy used})(\text{rate}) = (15 \text{ kWh})(\$0.080 / \text{kWh}) = \$1.20 \quad \diamond$$

Observe that this cost is independent of both the resistance of the heating coil and the applied voltage.
